101. Which electron dot formula represents a polar molecule?

⁽¹⁾ Ö	::C::Ö	(3)	н:О: Н
⁽²⁾	<u>і</u> Э:Н І	(4)	:ĊI: :ĊI:Ċ:ĊI: :ĊI:

102. Which molecule is nonpolar?

(2) Hg(*l*)

(1)	H ₂ O	(3)	СО
(2)	NH ₃	(4)	CO_2

103. Carbon dioxide is a colorless, odorless gas produced by respiration and combustion of carbon-containing fuels. It is slightly soluable in water.

Draw the Lewis electron-dot structure for a molecule of carbon dioxide, CO2.

(4) $H_2O(g)$

	pressure of 1 atmosphere, w ure of steam and water at eq				
(1)	100°C	(3)	273°C		
(2)	212°C	(4)	373°C		
105. A 1-gram sample of which substance in a sealed 1-liter container will occupy the container completely and uniformly?					
(1)	Ag(s)	(3)	$\mathrm{H_2O}(\ell)$		

- 106. Which sample is composed of particles arranged in a regular geometric pattern?
 - (1) $Cl_2(g)$ (3) LiCl(s)(2) $CCl_4(\ell)$ (4) LiCl(aq)